

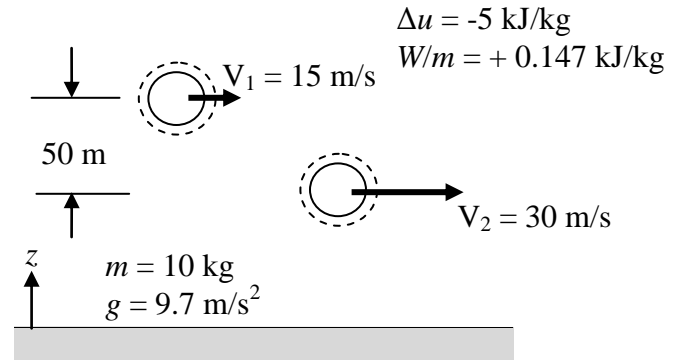
Problem 2.58

A closed system of mass of 10 kg undergoes a process during which there is energy transfer by work from the system of 0.147 kJ per kg, an elevation decrease of 50 m, and an increase in velocity from 15 m/s to 30 m/s. The specific internal energy decreases by 5 kJ/kg and the acceleration of gravity is constant at 9.7 m/s^2 . Determine the heat transfer for the process, in kJ.

KNOWN: Data are provided for a closed system undergoing a process involving work, heat transfer, change in elevation, and change in velocity.

FIND: Determine the heat transfer for the process.

SCHEMATIC AND GIVEN DATA:



ENGINEERING MODEL: (1) The system is a closed system. (2) The acceleration of gravity is constant.

ANALYSIS:

$$\Delta U + \Delta PE + \Delta KE = Q - W \quad \rightarrow \quad Q = \Delta U + \Delta PE + \Delta KE - W$$

$$\checkmark \quad W = m [W/m] = 10 \text{ kg} [-0.147 \text{ kJ/kg}] = -1.47 \text{ kJ}$$

$$\checkmark \quad \Delta U = m\Delta u = 10 \text{ kg} [-5 \text{ kJ/kg}] = -50 \text{ kJ}$$

$$\checkmark \quad \Delta KE = \frac{m}{2} (V_2^2 - V_1^2) = \frac{10 \text{ kg}}{2} \left[\left(30 \frac{\text{m}}{\text{s}} \right)^2 - \left(15 \frac{\text{m}}{\text{s}} \right)^2 \right] \left| \frac{1 \text{ N}}{1 \text{ kg} \cdot \text{m/s}^2} \right| \left| \frac{1 \text{ kJ}}{10^3 \text{ N} \cdot \text{m}} \right| = +3.38 \text{ kJ}$$

$$\checkmark \quad \Delta PE = mg(z_2 - z_1) = (10 \text{ kg}) (9.7 \text{ m/s}^2)(-50 \text{ m}) \left| \frac{1 \text{ N}}{1 \text{ kg} \cdot \text{m/s}^2} \right| \left| \frac{1 \text{ kJ}}{10^3 \text{ N} \cdot \text{m}} \right| = -4.85 \text{ kJ}$$

$$Q = (-50) + (-4.85) + (3.38) - (-1.47) = -50 \text{ kJ (out)} \quad \leftarrow$$